



WATSON WG3

Using environmental isotopes for waiting time assessment in groundwater protection zones

Julien Farlin, July 8th 2021



LE GOUVERNEMENT
DU GRAND-DUCHÉ DE LUXEMBOURG
Ministère du Développement durable
et des Infrastructures

Administration de la gestion de l'eau

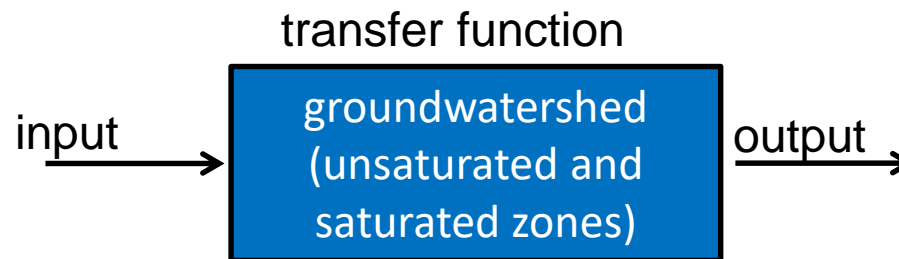


Estimates of waiting times are useful when mitigation measures are taken in a protection zone, and you want to know how long it will take for water quality to improve. For instance:

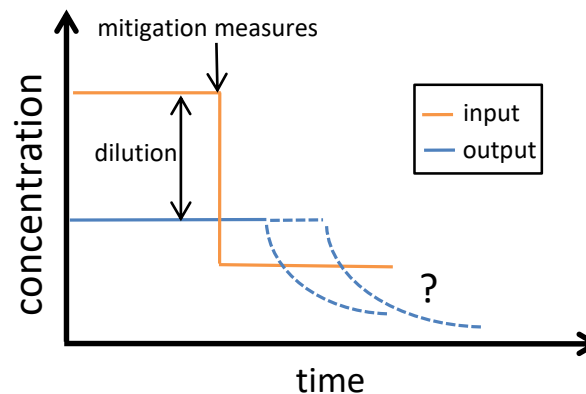
- A upward trend is to be reversed, but nothing happens. Are the measures insufficient, or is the aquifer acting as buffer ?
- An agrochemical has contaminated the groundwater. It is now forbidden, but how long will the water provider have to wait until groundwater can be used again for drinking purposes ?



If we frame the transport of agrochemicals from the fields to the groundwater intake (well or spring) as a classical input-output problem, we have



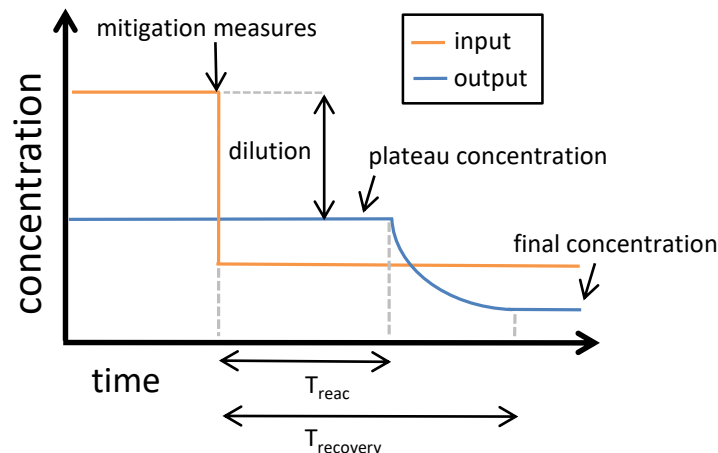
A water manager is interested in predicting the future output for certain mitigation scenarios

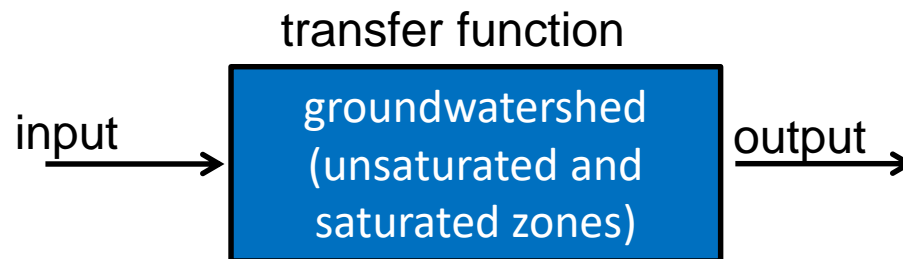


Some useful waiting times



- mean transit time \bar{T} : average travel time from injection point to outlet
- reaction time T_{reac} : lag between measure implementation and begin of change at the outlet
- recovery time $T_{recovery}$: lag between measure implementation and reaching a certain concentration goal (e.g. drinking water standard)
- Both T_{reac} and $T_{recovery}$ depend upon \bar{T} , but are not equal to it !





- input → estimated/reconstructed/simulated
- transfer function → calibrated with isotopes or historical contaminants
- output → sought information



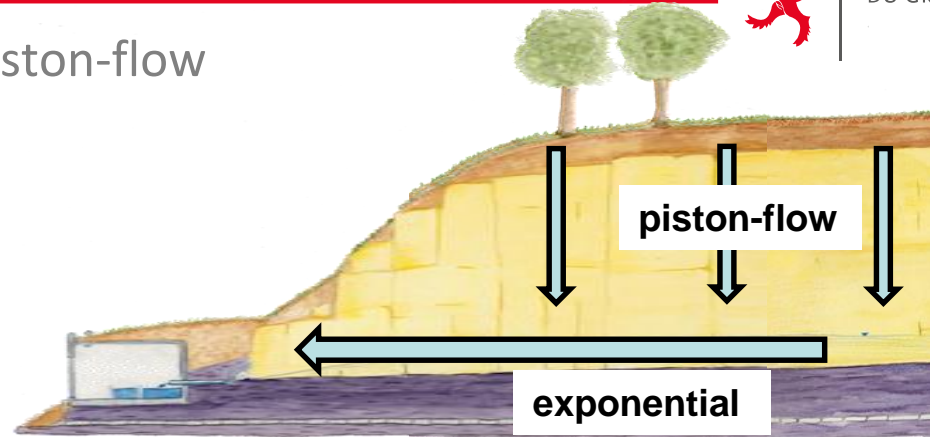
Several springs providing drinking water to the city of Luxembourg have been abandoned due to increasing concentrations of n,n-dimethylsulfamide (TP of the orchard fungicide tolylfluanide). Tolylfluanide was banned from use in 2008. The water provider wanted to know:

- What the maximum concentrations would be.
- When this maximum would be reached.
- The waiting time until concentrations would be back to the drinking water standard of 100 ng/L.

Case study

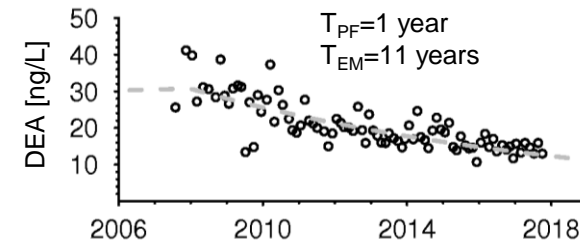
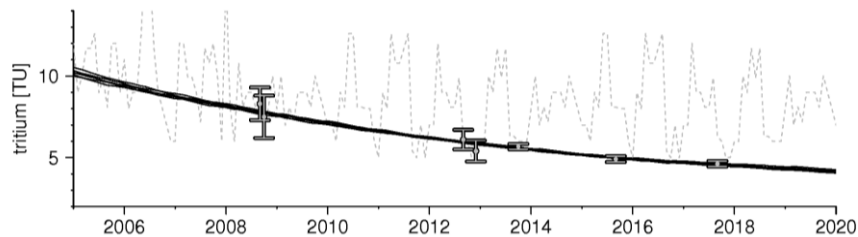


- Model: exponential piston-flow



- Calibration with $^3\text{H}+^2\text{H}$ and desethylatrazine

➔ $\bar{T} = T_{EM} \approx 11-15$ years

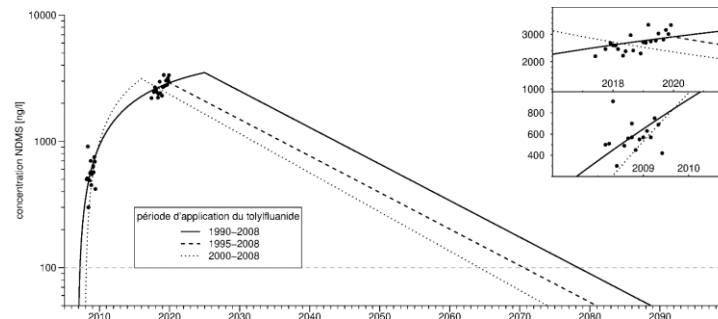


- Prediction

➔ $T_{reac} \approx 15$ years

➔ $T_{recovery} \approx 2070-2080$

➔ $C_{max} \approx 3000$ ng/L





- Classical groundwater dating methods are simple to adapt to (ground)water quality problems.
- The transfer function can be parameterized using environmental isotopes or historical contaminants (combining both is best)
- The different waiting times useful to water managers should not be confused with the tracer mean transit time.
- Useful waiting times can be *derived* from the calibrated transit time distribution.



Question ?